ACC NR: AP6023769	SOURCE CODE: YU/0015/65/000/02-/0069/0070
AUTHOR: Milosevic, Sava (Doc	ctor); Vidakovic, Dobrinka
ORG: Federal Institute of Pu (Savezni zavod za zdravstvenu	ublic Health/directed by Doctor H. Kraus/, Belgrade
TITLE: Medical center and it	ts tasks
SOURCE: Medicinski glasnik,	no. 2-3, 1965, 69-70
TOPIC TAGS: medical facility	
ARSTRACT. General review of	
legislative changes: status and institutional cooperation	organizational and administrative position and the center in Yugoslavia according to recent and relationship with various communal authorities is role in education and in treatment; monitoring ganizations involved of the Center functions.
legislative changes: status and institutional cooperation by the various persons and or JPRS]	and relationship with various communal authorities
legislative changes: status and institutional cooperation by the various persons and or JPRS]	and relationship with various communal authorities is role in education and in treatment; monitoring ganizations involved of the Center functions.
legislative changes: status and institutional cooperation by the various persons and or JPRS]	and relationship with various communal authorities is role in education and in treatment; monitoring ganizations involved of the Center functions.

KONSTANTINOVIC, Ivan; LICHT, Antun; VIDAKOVIC, Helena

Alcoholism in school children in the Subotica District; Higijena,
Beogr. 9 no.2-3:147-153 1957.

(ALCOHOLISM, in inf. & child
school child, in Yugosl. (Ser))

M. VIDAKOVIC

"A contribution to the identification of pinus halepensis Mill." p. 11 (SUMARSKI LIST, Vol. 77, no. 1, Jan. 1953, Zagreb, Yugoslavia)

SO: Monthly List of the East European Accessions, L.C., Vol. 2, No. 7, July 1953, Uncl.

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"

VIDAKOVIC, M.

VIDAKOVIC, M. Importance of anatomic structure of needles in species of black pine in Yugoslavia. p. 244.

Vol. 79, no. 7/8, July/ Aug 1955 SUMARSKI Zagreb, Yugoslavia

So: Eastern European Accession Vol. 5 No. 4 April 1956

VIDAKOVIC. H.

Genetics in forestry. p. 103 (GLASNIK, Vol. 80, No. 3/4, Mar./Apr. 1956

SO: Monthly List of East European Accessions (EEAL)LC Vol. 6, No. 12, Dec. 1957 Uncl.

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"

FIOLIC, V., inz.: VIDAKOVIC, R., inz.

Some observations on the use of Tekstofix in textile industries. Kemija u industrijī no 5:301-303 My '62.

Commence of the second second second

FIOLIC, V., inz.; VIDAKOVIC, R., inz.

Problem of freeing plywood sheets from formaldehyde. Kem ind 13 no. 2: 125-128 F '64.

VIDAKOVIC, R., inz.

Assortment of synthetic resins for wood at the Chromos Factory, Zagreb. Kem ind 13 no. 2:129-133 F 164.

VIDAKOVIC, S.

Structure of multistoried dwellings and urban municipal rervices and analytic study. p. 687. TEHNIKA (Savaz inzenjera i techicara Jugoslavije) Beograd. Vol. 11, no. 5, 1956

SOURCE: East Europe Accessions Lists (EEAL), Library of Congress, Vol. 5, no. 11, Nov. 1956

WINAKOVIC S

YUGOSLAVIA/Tumore

U-4

Abs Jour : Ref Zhur - Biol., No 6, 1958, No 27912

Author : Videkovic, S., Dekaris, M., Grgurevic, M.

Inst : Not Given

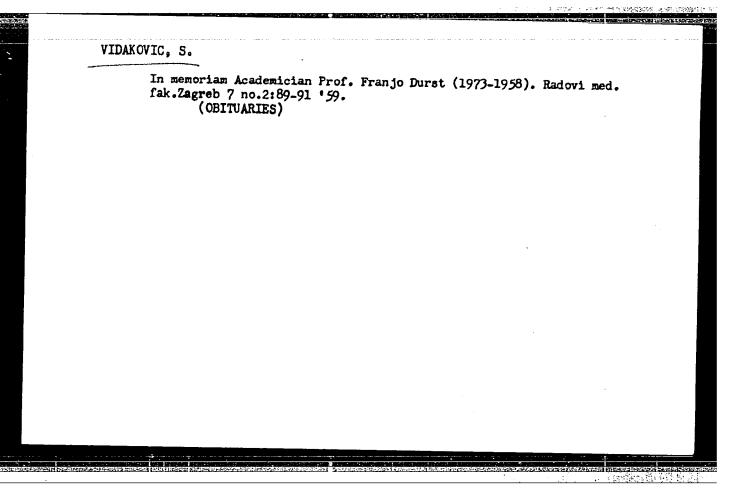
Title : The Treatment of Corcinena of the Cervix

Orig Pub: Liljocn. vjesn., 1956, 78, No 7-8, 316-330

Abstract: 1437 patients with carcinena of the cervix were treated in bynecologic clinics at Zegreb. Of 603 patients (stages I and II) 241 females (39.9%) were subjected to various methods of surgical treatment. 71.4% were cured. Surgical intervention was resorted to in 14 patients, with stages III and IV, following irradiation. Postoperative death rate was 1.5%. Severe complications were observed in 1u patients. X-ray therapy was used on 1182 patients. Positive five-year results for all stages comprised 30.5%. The overall cure rate was 37.3%. Histological diagnosis of adencearcinena does not require surgical treatment. The authors believe that the best treatment is a combination of surgery and irradiation.

Card : 1/1

43



VIDAKOVIC, Stjepan; BAGOVIC, Pero; DRAZANCIC, Ante; RAPIC, Smail

Physiology of uterine contractions and its measurement. Radovi med. fak., Zagreb 7 no.3:221-232 *59.

(UTERUS physiol)

VIDAKOVIC, Teodor, inz.

Automatic control of the operational output of machines.
Tehnicki pregled 13 no.4:151-156 '61.

VIDAKOVIC, Teodor, inz.

Selecting the most economical combustible for the needs of heat energy in industry. Tehnicki pregled 14 no.6:223-226 '62.

VIDAKOvIC, Velibor, inz., saradnik (Beograd, Jevrema Grujica 14); MILOJEVIC, Mihajlo, ekon., saradnik

A plan for the development of urban traffic in Serbia. up to 1970. Tehnika Jug 19 no.6: Suppl: Hemindustrija 18 no.6:1143-1150 Je '64.

1. Belgrade Urbanization Office, Belgrade.

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"

SKARICA, Radoslav, dr.; VIDAKOVIC, Zdenko, dr.; KADRNKA, Rajka, dr.

Significance of Ortolani's position in early roentgenological diagnosis of congenital hip dislocation. Lijec. vjes. vjes. 81 no.11:819-825 '59.

1. Iz Zavoda sa rentgenologiju i Odjela sa djecje bolesti bolesti Opce bolnice "Dra. M. Stojanovica" u Zagrebu. (HIP fract. & disloc.)

VIDAKOVIC, Zdenko, Dr.; BALOG, Marijan, dr.

Use of water-soluble and isotonic propyliodone in bronchography. Lijec. vjes. 78 no.11-12:546-549 Nov-Dec 56.

1. Iz Zavoda za rentgenologiju i Odjela za uho, nos i grlo Opce bolnice dr. Stojanovica u Zagrebu. (CONTRAST MEDIA,

water-soluble & isotonic propyliodone in bronchography (Ser))

(BRONCHI, radiography

water-soluble & isotonic propyliodone contrast medium (Ser))

SOV/86-58-8-20/37

AUTHOR:

Vidakas, Yu.E., Engr Maj

TITLE:

Adjustment of Radar Sights (Yustirovka radiolokatsion-

nykh pritselov)

PERIODICAL:

Vestnik vozdushnogo flota, 1958, Nr 8, pp 56-58 (USSR)

ABSTRACT:

The article describes the adjustment of radar sights by a method suggested by an officer of an air force unit, A.T. Nikolayev. By this method the adjustment of sights is made with the aid of an "artificial target" which, after some modifications, can be successfully used instead of the radar tester and the pulse generator. The device "artificial target" is a pulse generator of high frequency, operating on the same frequency as the radar sight to be adjusted, and is triggered by the high frequency signals. At each pulse of the sight the device responds with one high frequency pulse of the same duration, but with some delay. Thus

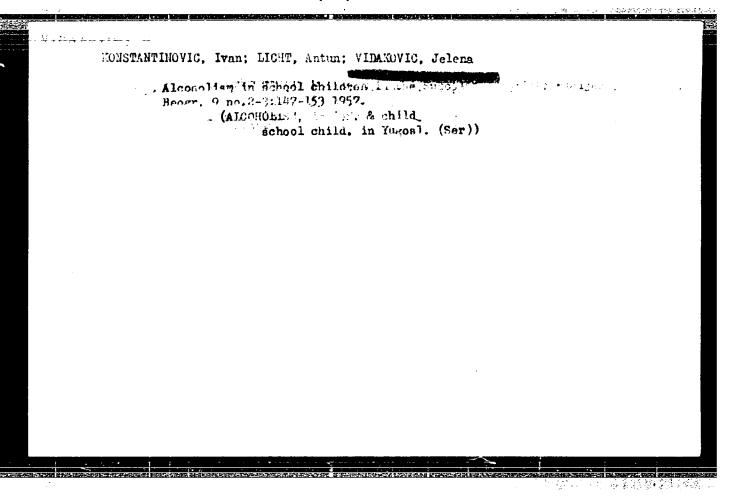
Card 1/2

Adjustment of Radar Sights

sov/86-58-8-20/37

the device simulates the real targets at any given distance. The accuracy of adjustment by this method, according to the author, is sufficiently high. Three diagrams.

Card 2/2



VIDAKOVIC, V.; BEROVIC, D.

Armored brigade in defensive maneuver. p. 26.

VOJNI GLASNIK. (Jugoslavenska narodna armija) Beograd, Yugoslavia Vol. 9, no. 8, Aug. 1955

Monthly List of East European Accessions (EEAI) LC, Vol. 8, no. 9, Sept. 1959.

Uncl.

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LIHT, Antun, Dr.; VIDAKOVIE, Jelena, lek. Pomocnik

Fungi from damp dwellings as a cause of allergy. Higijena,
Beogr. 8 no.2-3:212-213 1956.

1. Higijenski zavod u Subotici.
(ALLEROY, etiol. & pathogen.
fungi in damp dwellings (Ser))
(FUNGUS DISEASES, etiol. & pathogen
allergy from fungi in damp dwellings (Ser))
```

ACCESSION NR: AT4022941

P/2536/63/000/004/0003/0026

AUTHOR: Vidal, Pierre

TITLE: Analysis of stability of nonlinear sampled data control

SOURCE: Oliwice. Politechnika Slacka. Zeszyty naukowe, no. 90, 1963. Automatyka (Automation) no. 4, 3-26

TOPIC TAGS: control system, sampled data control system, nonlinear control system, automatic control system, control system stability, pulse width modulation, control system analysis, voltage stabilizer

ABSTRACT: Paper presents an analysis of an industrial voltage stabilizing system whose operation has a nonlinear sampled data nature. The stability of a nonlinear sampled data control system with pulse width modulation is discussed. Author attempted to find asymptotic stability conditions for a system containing controlled rectifiers. The area of the asymptotic stability show in Ol of the Enclosure can be defined from the equations:

Card 1/5

ACCESSION NR: AT4022941

$$E_2 = \frac{2T}{k} \left(\frac{T}{k} \frac{1+D}{AD} - 1 \right)$$
 (1)

and

$$\varepsilon > \frac{AD}{1+D} \left(e^{-(+)} \frac{T}{T} - 1 \right) \tag{2}$$

The amplification k, defined by equation (1), permits a computation of the maximum step \mathcal{E}_2 which is permissible with respect to stability. The

instability area is contained between two asymptotic stability areas. All points found in this are yield non-zero or fluctuating erros, and do not have asymptotic stability. Author also discusses a sampled data system with pulse width modulation and dynamic nonlinearity. The asymptotic stability area is defined by the inequalities

Card 2/64

ACCESSION NR: AT4022941

$$\begin{cases} \varepsilon < \varepsilon'_{2} & \frac{k'}{T} \varepsilon'_{2} = \log \left(1 + \varepsilon'_{2} \frac{1+D}{AD}\right) \\ \varepsilon > \varepsilon_{0} & \frac{T_{0}}{1+D} \left(e^{\alpha'_{2} \left(\frac{1}{T}\right)} - 1\right) \end{cases}$$
(3)

These findings are then used to design a nonlinear correction element with which an optimal operation of the entire control system can be achieved. The introduction of the nonlinear element brought about a higher accuracy in the control of the output voltage under steady state conditions. On the other hand, there is greater chance of larger disturbances in the open circuit, owing to which the system responds very quickly to disturbances. The nonlinear element made it possible to obtain maximum (optimal) parameters of the dynamic state which were in harmony with the stability conditions. "The present work was carried out at the Department of Control Theory of the Silesian Polytechnic under the direction of Professor Doctor Stefan Wegrzyn. Author wishes to thank him for his constant help and advice. Author also wishes to thank the personnel of the Department of Control Theory."

Card 3/5/-/

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Card 4/8 1	4/6			•	

WEGRZYN,S., VIDAL,P.

Asymptotic stability of gauged systems. Bul Ac Pol tech 11 no.2:99-101 *63.

1. Chaire de la Theorie de Regulation, Institut Polytechnique de Silesie, Gliwice, et Laboratoire de Genie Electrique, Universite de Toulouse, France. Presented by J. Groszkowski.

VIDALI, Primoz

A case of toxic reaction of the bone marrow to cytostatic drugs treated by bone marrow transplantation. Zdrav. vestn. 33 no.6: 178 '64

1. Interni oddelek splosne bolnisnice Jesenice (Predstojnik: prim.dr. Fr. Brandstetter).

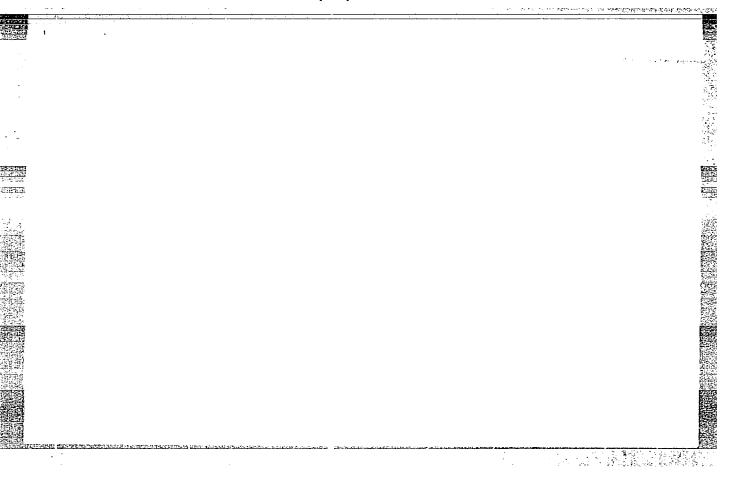
MEGRZYN, S.; VIDAL, P.

Certain Liapunov function and the asymptotic stability of gauged systems. Bul Ac Pol tech 11 no.1:45-51 *63.

1. Chaire de Theorie de Regulation, Universite Technique de Silesie, Gliwice, et Laboratoire de Genie Klectrique, Universite de Toulouse, France. Presented by J.Groszkowski.

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NASTAGE, M. Gr.; VIDAN, M.

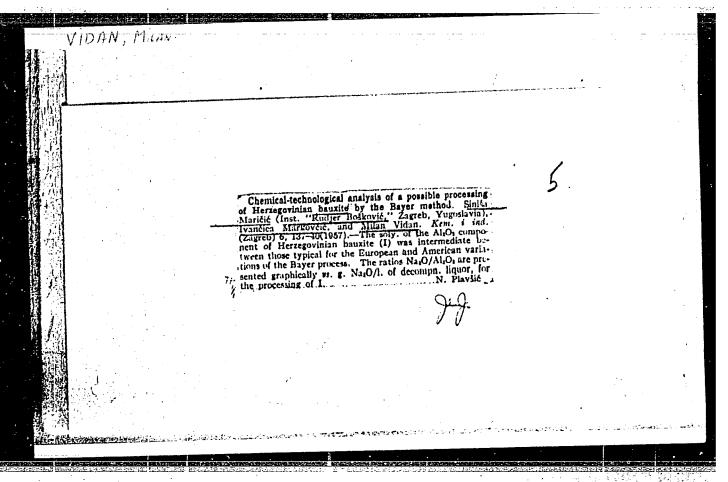
Variation of the absorption coefficients of X rays, depending on the thickness of the absorbing material. Rev mec appl 9 no.5:1101-1114 164.

1. Institute of Applied Mechanics, Bucharest.

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"

The kinetics of formation of aluminum hydreride by seeding sadium aluminate polations with hydrogralities granta. III. The effect of coarse-grained seed. S. Maraké and M. Vidan (Iust. Lake Metale, Zagreb Yugo-Kineti, Archivery 17, 215-16(1950) in English; cf. C.A 49, 1537a.—Decomputates and particle-size cumulative distribution curves, obtained by pptg. Alch.214.0 from Neudaminus when, by excling with coarse-grained Alch.314.0 crystals, give further evidence that nucleation is due to fine particles independently of the coarse times. N. Plavšić

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"



B-8

VIDAN.M. YUGOSLAVIA/Physical Chemistry - Thermodynamics,

Thermochemistry, Equilibria, Physical-Chemical

Analysis, Phase Transitions.

: Referat Zhur - Khimiya, No 1, 1958, 390 Abs Jour

: P. Bogdanovic, S. Maricic, M. Vidan. Author

Inst

: The Kinetics of The Formation of Aluminium Hydroxide by Seeding Sodium Aluminate Solutions with Hydrar illite Title

Crystals. IV. On the "Contact-Intercrystallization."

: Croat. chem. acta, 1956, 28, No 3, 155-162 Orig Pub

: It is shown that the growth of Al(OH)3 grains in solutions Abstract

of Na aluminate deposited by hydrargillite is caused by merging of separate little particles, which has been called "the contact intercrystallization". A method of choosing among various crystallization mechanisms is proposed.

Card 1/2

CIA-RDP86-00513R001859710014-5" APPROVED FOR RELEASE: 09/01/2001

YUGOSLAVIA/Physical Chemistry - Thermodynamics,

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B-8

计对称 推断数学

Thermochemistry, Equilibria, Physical-Chemical

Analysis, Phase Transitions.

Abs Jour

: Ref Zhur - Khimiya, No 1, 1958, 390

It is based on the study of the dependence of the ratio between the final number of particles and the initial on time. Also, some proposals concerning the practical application are brought forward. The most convenient method of computation of the distribution of particle weights according to their dimensions is discussed. See part III in RZhKhim, 1957, 25983.

Card 2/2

VIDAN, MILAN

YUGOSLAVIA/Chemical Technology - Chemical Products and Their

H-8

Application - Elements, Oxides, Mineral Acids,

Bases, Salts.

Abs Jour

: Ref Zhur - Khimiya, No 3, 1958, 8603

Author Inst

Maricic Sinisa, Markovcic Iwancica, Vidan Milan

Title

: Technological Study of the Possibility of Processing Herzegovina Bauxites by the Bayer Method. I. Chemical Laboratory Method of Bauxite Classification and Planning

of Industrial Scale Experiments.

Orig Pub

: Kemija u industriji, 1957, 6, No 5, 137-140

Abstract

: No abstract.

Card 1/1

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"

YUGOSLAVIA/Chemical Tochnology: Chemical Products H-8 and Their Applications. Elements. Oxi-

des. Mineral Acids. Bases. Salts.

Abs Jour: Ref Zhur-Khimiya, No 7, 1959, 23964

Author : Vidan, M.

Title : Effect of Mixing on the Decantation Process

Involving Red Residue Obtained in the Refi-

ning of Herzegovina Bauxites.

Orig Pub: Kemija u industriji, 1958, 7, No 5, 117-120

Abstract: It is established experimentally that the

behavior of red residue (formed in the manufacture of Al₂0₃ in accordance with the Bayer method) in the decantation depends on agitation of the reaction mixture during

Card : 1/2

YUGOSLAVIA/Chemical Technology. Chemical Products H-8 and Their Applications. Elements. Oxides. Mineral Acids. Bases. Salts.

Abs Jour: Ref Zhur-Khimiya, No 7, 1959, 23964

the alkalization stage. The degree of particle aggregation is affected by coagulating agents, however, the end result depends largely on the mixing of pulp. An intense mixing irreversibly destroys structure of the residue, while aggregates, formed under the influence of coagulating agents, have lower density.

Card : 2/2

14-39

NASTASE, M. Gr.; VIDAN, M.

Variation of coefficients of absorption of I rays depending on the thickness of absorbing material. Studii cerc mec apl 16 [i.e. 15] no.3:729-744 '64.

1. Submitted February 21, 1964.

Proizvodstvo *uchnykh konditerskikh izdelly (Production of farinaceous confectionary, by) K. Kh. Vidanov i S. A. Kotel'nikov. Moskva, Fishcherromizdat, 1953.
207 p. illus., diagrs., tables
Literatura: p. 208

SO: N/5
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described in the second of the

VIDELLY, K. Kn.

"Investigation in the Field of Technological Combination and the Typical Plan ing of Enterprises in the Food Industry." Cand Technological Losingrad Technological Inst of the Refrigeration Industry, Min Higher Education USSR, Leningrad, 1954. (NL, No 2, Jun 55)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions. (13) SO: Sum. No. 598, 29 Jul 55

285h Videnov, K. Kh.

Issledovanie v oblasti tekhnologicheskogo kombinirovaniya i tapovogo procekhtirovaniya preduriyatiy pishevoy promyshlennosti. (na primere proizvodstv muchnykh izdeliy shirokogo assortimenta). L., 1954. 15 s. s graf. 20 sm. (M-vo vyssh. obrazovaniya SSSR. Leningr. tekhnol. in-t kholodil-noy prom-sti). 100 Ekz. B. ts. - (54-56163)

VIDANOV, K. Kh.

"Investigation in the Field of Technological Combination and the Typical Planning of Enterprises in the Food Industry." Cand Tech Sci. Leningrad Technological Inst of the Refrigeration Industry, Min Eigher Education USSR, Leningrad, 1954. (KL. No 2, Jan 55)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (12)

SO: SUM No. 556, 24 Jun 55

VIDANOV, K.Kh.; KOTEL'NIKOV, S.A.; KAIMENS, R.I., redaktor; DUBOVKINA,

N.A., tekhnicheskiy redaktor

[Bakery goods production] Proizvodstvo muchnykh konditerskikh
izdelii. Meskva, Pishchepromizdat. 1953. 207 p. (MIRA 7:9)

(Baking)

MARKHEL', Pavel Sil'vestrovich, kand. tekha.mauk; SMELOV, Sergey Vasil'yevich, master-konditer; MASLOV, Ivan Nikolayevich, kand. tekhn. nauk; DANILEVSKAYA, Valentina Vladimirovna, kand. tekhn. nauk; GOPENSHTEYN, Yuriy Lazarevich, inzh.; VIDANOV, Konstantin Kharitonovich, inzh.; ZAPENINA, Nina Vasil'yevna, kand. tekhn. nauk; SOKOLOVA, Nina Ivanovna, tekhnolog; PRITYKINA, L.A., red.; KISINA, Ye.I., tekhn.red.

[Confectionery products made with flour] Muchnye konditerskie izdeliia. [By] P.S.Markhel i dr. Moskva, Pishchepromizdat. Pt.l. [Making of pastries, torten, cakes, oriental and dietetic products] Proizvodstvo pirozhnykh, tortov, keksov, vostochnykh i dieticheskikh izdelii. 1962. 679 p. (MIRA 16:7)

(Baked products)

VIDANOV, Konstantin Kharitonovich; TOKAREV, L.I., kand. tekhn.nauk, spets. red.; PRITYKINA, L.A., red.; KISINA, Ye.I., tekhn. red.

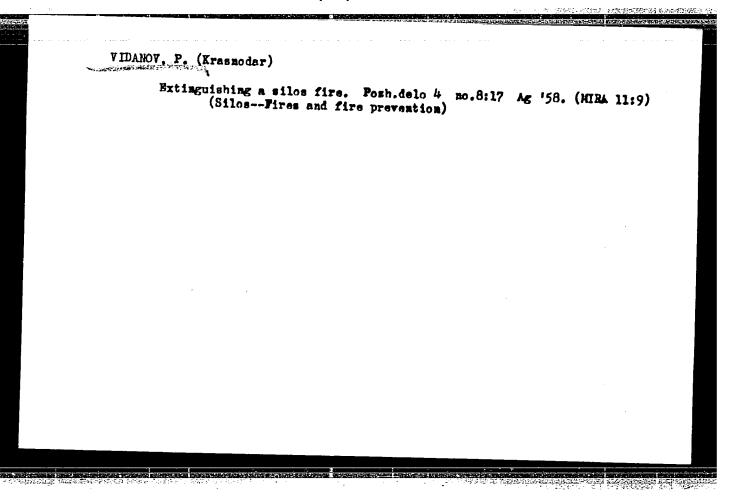
[Baked products] Muchnye konditerskie izdeliia. Moskva, Pishchepromizdat. Pt.2.[Production of cookies, bisquits, crackers, gingerbread and waffles] Proizvodstvo pechenia, galet, krekerov, prianikov i vafeli. 1962. 233 p.

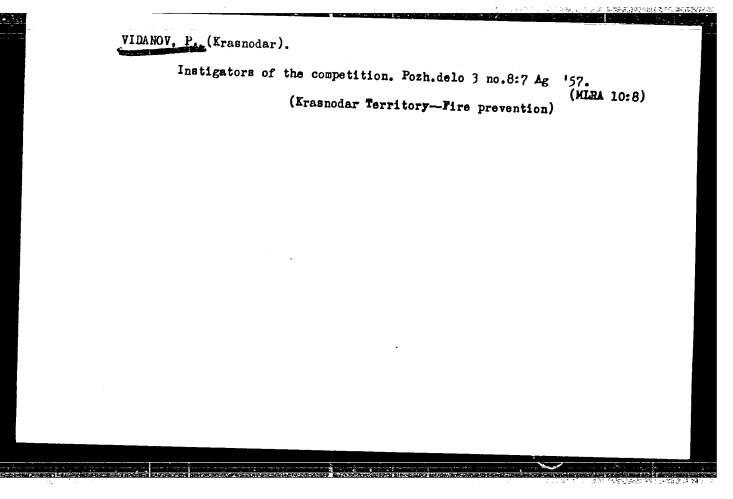
(MIRA 15:8)

(Baked products)

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SINYAKOV, Aleksandr Borisovich; ANTIPOVA, Anisiya Ivanovna; KARASEVA, Nina Nikolayevna; AVER'YANOVA, T.N., inzh., retsenzent: VIDANOVA, R.I., prepodav., retsenzent; GUR'YANOVA, N.I., prepodav., retsenzent; DATNER, M.G., inzh., retsenzent; KARASEV, V.K., kand. tekhn. nauk, nauchn. red.; GABOVA, D.M., red.

[Technology of clothing manufacture] Tekhnologiia shveinogo proizvodstva. Moskva, Legkaia industriia, 1965. 409 p. (MIRA 18:7)

VIDANOV. V.A. KASHEKOV, L.Ya., inzhener; YAKUSHUNKOV, S.M., inzhener; MATVEYEVA, Ye.N., tekhnicheskiy redektor

[BN-3 drilling and pumping unit developed by V.A. Vidanov] Buril'no-nasosnaia ustanovka BN-3 sistemy V.A. Vidanova. Moskva, Gos. nauchnotekhn. izd-vo mashinostroit. lit-ry. 1957. 21 p. (MIRA 10:7)

1. Vsesoyuznyy institut mekhanizatsii (for Kashekov, Yakushenkov) (Boring machinery) (Pumping Machinery)

VIDANOVIC-SAZDA, GAVRILO

Visok; privredno-geografska inspitivanja. Beograd, Naucna knjiga, 1955. 195 p. (Srpska Akademija nauka. Posebna izdanja, knu. 238)

SOURCE: Bast European Accessions List, (EEAL) Library of Congress, Vol. 5, No. 8, August, 1956.

VIDAR, M.

roblems of glass packaging in our canning industry with special reference to the system of sealing and proposals for improving existing conditions. p. 1229. TEHMKA (Savaza inzenjera i techicara Jugoslavije) Beogard. Vol. 11, no. 8, 1956

SOURCE: East Europe Accession List (EEAL), Library of Congress, Vol. 5, no. 11, Nov. 1956

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001859710014-5"

VIDAS-POSEDEL, Zdenka

Pag; a contribution to the knowledge of insular settlements. Geogr glas 22:67-78 '60 (publ '61).

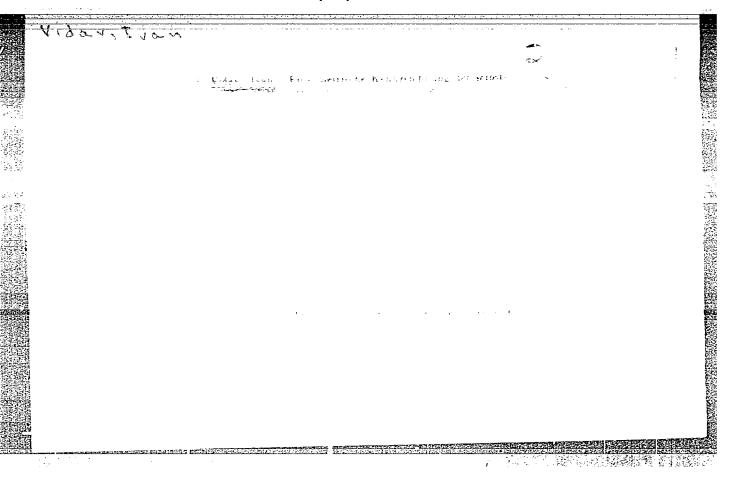
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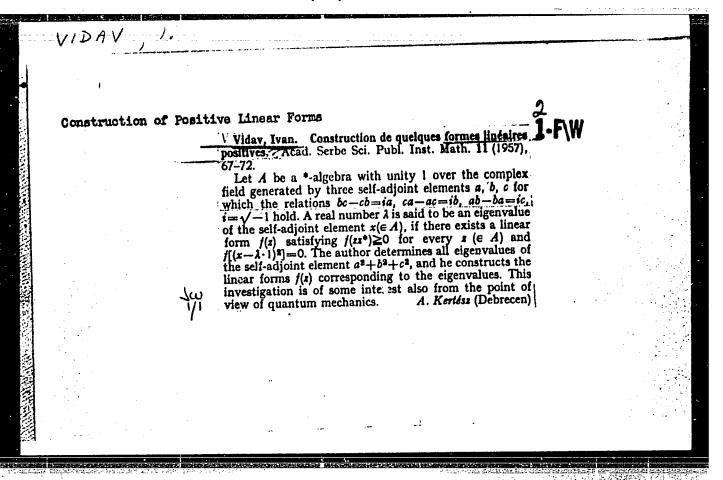
KOZLOV, I.; LYSOV, P. (Syktyvkar); VIDASHCHENKO, N., slesar'

Readers report. Sow. profsoluzy 19 no.1:20-21 Ja '63. (MIRA 16:1)

1. Neshtatnyy korrespondent zhurnala "Sovetskiye profsoyuzy", Bukhara (for Kozlov). 2. Khar'kovskiy zavod torgovogo mashinostroyeniya (for Vidashchenko).

(Labor and laboring classes)

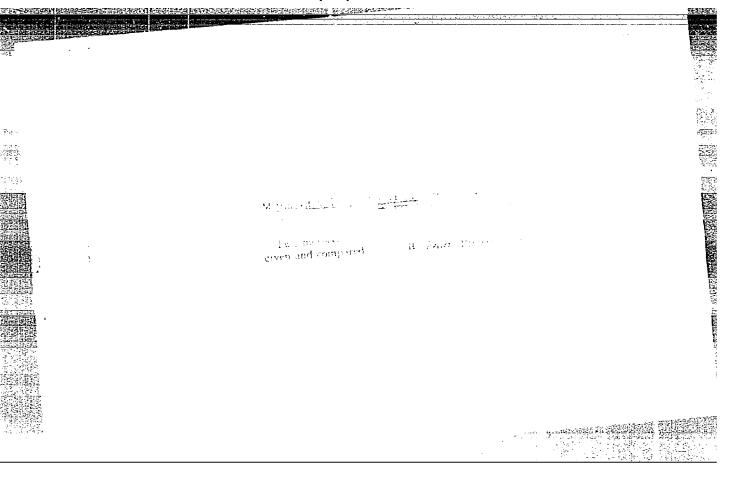


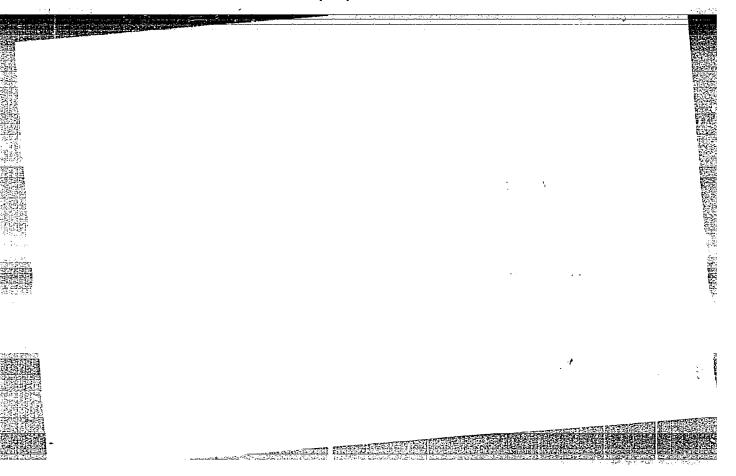


VIDAY, IVAN.

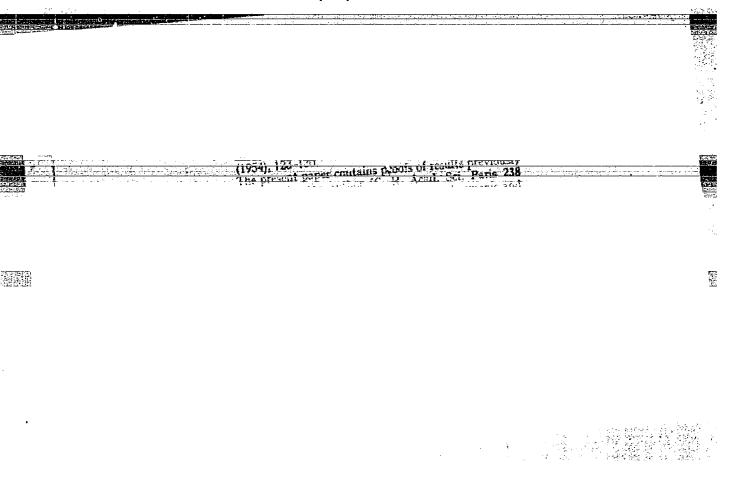
Vidav, Ivan. Visja matematika. Ljubljana (Drzavna zalozba Slovenije) 1949. Vol. 1. (Higher mathematics)

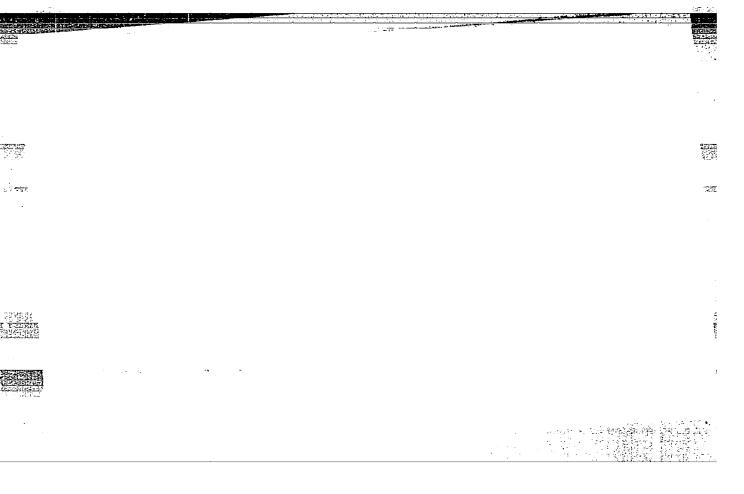
30: Monthly List Of East European Accessions, LC, Vol.3, No.1, Jan. 1954, Uncl.

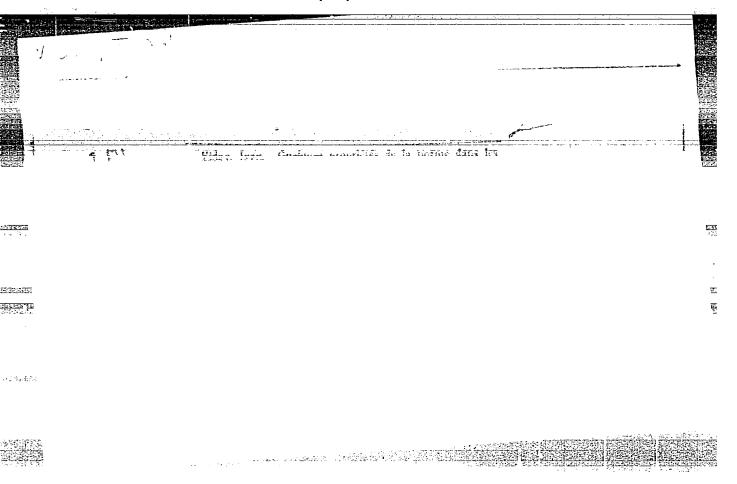




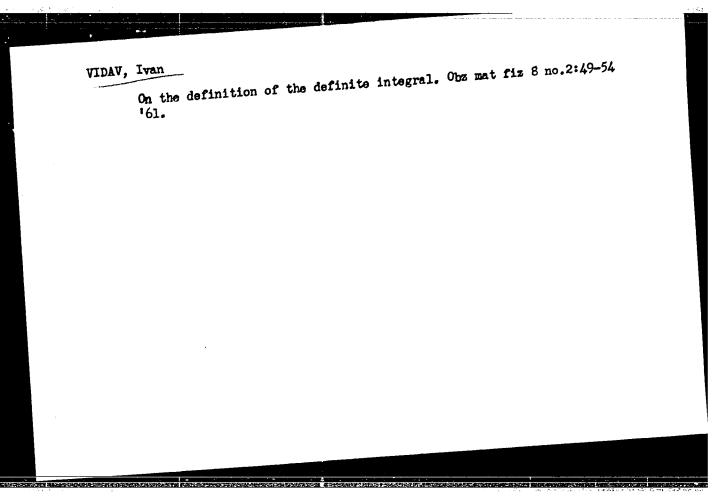
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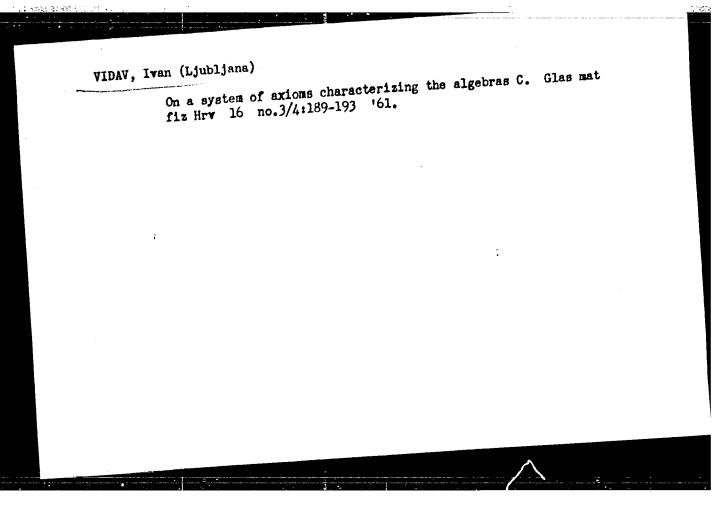




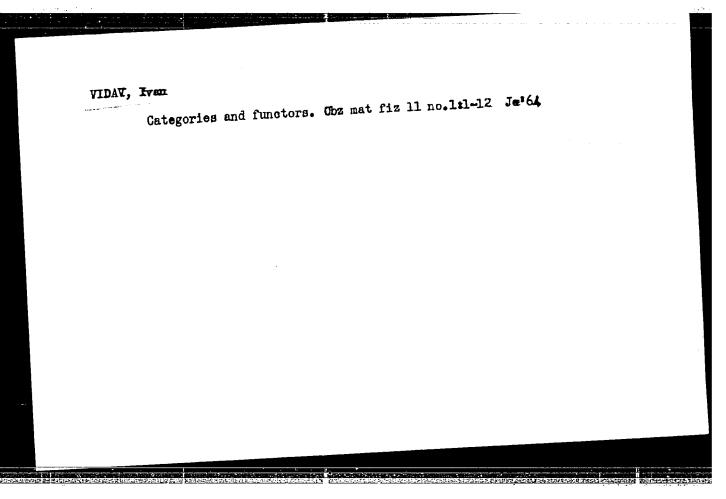


l. Drustvo matematikov in fizikov LRS; Urednistvo, Obzornik za l. Drustvo matematiko in fiziko. matematiko in fiziko. (Stojakovic, M.D.) (Maps) (Mathematics)	VIDAV, I	.A.Dirac and M.D.Stojako	ovic's Problem cetiri b ew. Obz mat fiz 7 no.48	(EEAI 10:5)
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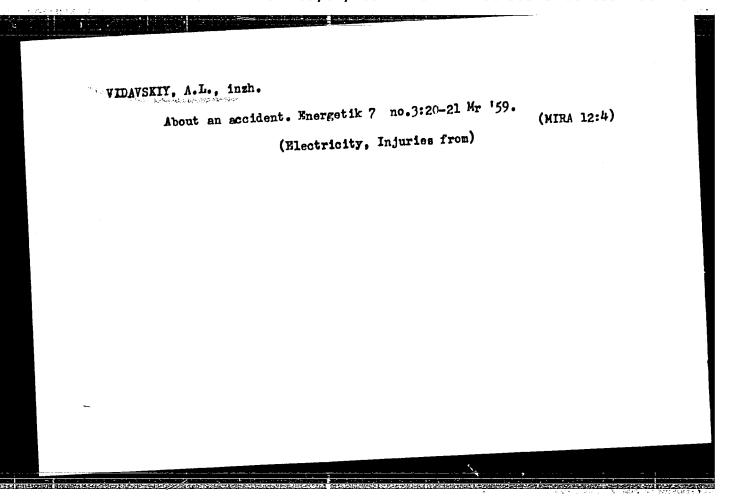
VIDAV, I. Head Professor Janko Branc; obituary. Obz met fix 10 no.3: 136 Nº63. 1. Clan Uredniskeg odbora, "Obzornik za matematiko in fiziko".



PODOL'SKIY, M.V., VIDAVSKAYA, G.M.

Study of the microstructure of dry blood preparations. Problegemat. i perel. krovi no.5:51-53 '65. (MIRA 18:10)

1. TSentral'nyy ordena Lenina institut gematologii i perelivaniya krovi (dir.-dotsent A.Ye. Kiselev) Ministerstva zdravookhraneniya SSSR, Moskva.



507/91-59-3-11/22

8(3,4)

Vidavskiy, A.L., Engineer AUTHOR:

TITLE:

A Fatal Accident (Ob odnom neschastnom sluchaye)

PERIODICAL:

Energetik, 1959,

Nr 3, pp 20-21 (USSR)

ABSTRACT:

The author describes a fatal accident in which an electrician was struck by current from a 400 volt line, believed to have no electric potential. The cause of this accident was a burned-out cable grounding the generator and a ground fault from one of the phases. There is l circuit.

Card 1/1

3/081/62/000/010/018/085 B138/B101

AUTHORS:

Vidavskiy, L. M., Kovba, L. M., Ippolitova, Ye. A.,

Spitsyn, Vikt. I.

TITLE:

Reaction of uranoso-uranic oxide with sodium and potassium

nitrates

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 93, abstract 10016 (Sb. "Issled. v obl. khimii urana". M., Mosk. un-t,

1961, 65 - 66)

TEXT: Using the methods of X-ray phase and thermal analysis it has been found that reaction between U₃O₈ and NaNO₃ begins at 410°C. As a result of this reaction the Na di-uranate is formed which reacts at a higher temperature (530°C) with the nitrate, to form the Na mono-uranate. As a result of interaction between the U₃O₈ and KNO₃ (beginning at 390°C) the potassium di-uranate is formed. (Abstracter's note: Complete translation.

Card 1/1.

\$/081/62/000/010/017/085 B138/B101

UTHORS: Vidavskiy, L. M., Kovba, L. M., Ippolitova, Ye. A.

AUTHORS: Vidavskiy, b. ...

Vidavskiy, b. ...

Vidavskiy, b. ...

Interaction between uranoso-uranic oxide and the sulfates of sodium and potansium

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 10, 1962, 92 - 93, abstract 10715 (Sb. "Issled. v obl. khimii urana". M., Mosk. un-t, 1961, 63 - 64)

TEXT: Using the methods of thermal and X-ray phase analysis, studies have been made of the reaction of U₃O₈ with Na and K sulfates. The reaction between U₃O₈ and Na₂SO₄ begins at 500°C. As a result of this reaction sodium di-uranate and UO₂SO₄ are formed which enter into reaction at a higher temperature, resulting in the formation of the di-uranate. The rehigher temperature, and U₃O₈, which begins at 580°C, is accompanied by action between K₂SO₄ and U₃O₈, which begins at 580°C, is accompanied by the formation of the potassium tri-uranate and UO₂SO₄. When the temperature is raised both these products react with K₂SO₄ to form the di-uranate Card 1/2

S/081/62/000/010/017/085
Interaction between uranoso-uranic ... B136/B101

[Abstracter's note: Complete translation.]

KOVBA, L.M.; VIDAVSKIY, L.M.; LAVUT, E.G.

Study of E -U03. Zhur.strukt, khim. 4 no.4:627-629 Jl-Ag '63.

(ETA 16:9)

1. Meskovskiy gosudarstvennyy universitet imeni Lemenoseva.

(Uranium exide orystals)

VIDAVSKIY, L.M.; KOVALICHUK, V. Yu.; BYAKHOVA, N.I.; IPPOLITOVA, Ye.A.

Enthalpy of oxidation with hydrogen peroxide of uranium (IV) sulfate terahydrate and octahydrate. Zhur. neorg. khim. 9 (MIRA 17:8) no.6:1489-1491 Je *63

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova, kafedra neorganicheskoy khimii.

VIDAVSKIY, L.M.; KOVAL'CHUK, V.Yu.; BYAKHOVA, N.I.; IPPOLITOVA, Ye.A.

Enthalpy of the reaction of amorphous uranium trioxide with
sulfuric acid. Vest. Mosk. un. Ser. 2: Khim. 19 no.5:65-68
(MIRA 17:11)

1. Kafecra neorganicheskoy khimii Moskovskogo universiteta.

VIDAVSKIY, L.M.; LAVUT, E.G.; KOVBA, L.M.; IPPOLITOVA, Ye.A.

Conditions of the formation of various modifications of uranium trioxide.

Dokl. AN SSSR 154 no.6:1371-1373 F '64. (MIRA 17:2)

1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. Predstavleno akademikom V.I.Spitsynym.

LAVUT, E.G.; VIDAVSKIY, L.M.

Autoclave for studying reactions over a wide range of temperatures.

Zhur. fiz. khim. 39 nc.21519-520 F '65. (MIRA 18:4)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova, khimicheskiy fakul'tet.

VIDAVSKIY, L.M.; BYAKHOVA, N.I.; IPPOLITOVA, Ye.A.

Enthalpy of the reaction of γ -UO3 with hydrofluoric acid and the enthalpy of γ -UO3 formation. Zhur. neorg. khim. 10 no.7:1746-1747 J1 '65. (MIRA 18:8)

1. Kafedra neorganicheskoy khimii Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova.

VIDAVEKIY, L.M.; BYAKHOVA, M.I.; KOVALTOHUK, V.Yu.; IPIOLITOVA, Ye.A.

Preparation of anorphous granies trioxide by thermal decomposition of granium percepted d hydrate. Vest. Mosk. un. Ser. 2: Alin. 19 no. 4:33-34 Ji-Ag 164. (MIRA 18:8)

1. Kafedra neorganich-skoy Phisii Moskovskogo universiteta.

VIDE, Albert, dipl. inz.

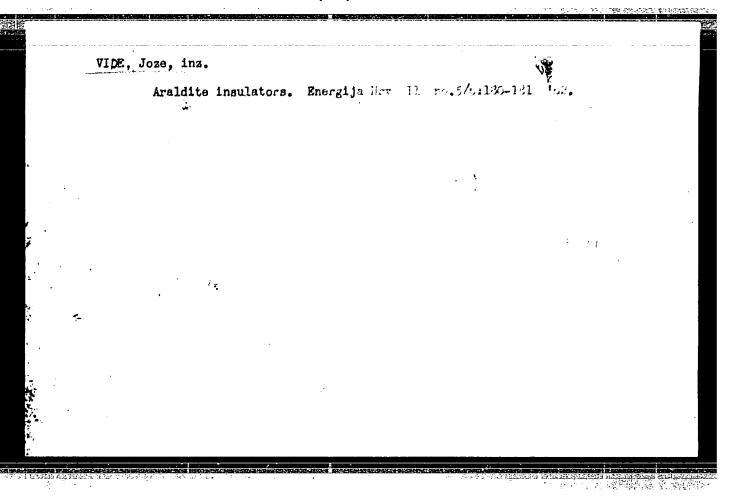
Applied theory of drives in mechanical engineering. Pt.1. Stroj vest 10 no.3:91-92 Je 164.

1. Institute of Motors and Motor Vehicles, Maribor.

VIDE, Albert, dipl. inz.

Applied theory of drives in mechanical engineering. Pt.2. Stroj vest 10 no.4/5:156 0 '64.

1. Institute of Motors and Motor Vehicles, Maribor.



VIDE, Joze, ing.

Behavior of armored, externally mounted transformer stations in Addis Ababa. Energija Hrv 10 no. 5/6:195-197 '61.

VIDE, Joze, ing. (Maribor)

Copper and aluminum in the construction of transmission lines. Energija Hrv 10 no. 7/8:243-246 '61.

1. Hidromontaza, Maribor.

VIDEANU, C., ing.

Applying the chain method. Constr Buc 16 no.745:1 18 Ap 164.

1. Seful santierului nr.3 Rimnicu-Vilcea al Trustului Regional de Constructii de Locuinte, Arges.

KUKOLJA, S. (Zagreb); POLAK, Lj. (Zagreb); KRNJEVIC, H. (Zagreb); VIDEK, M. (Zagreb)

Substances acting on the central nervous system. Croat chem acta 33 no.3:121-126 '61.

1. Research Department, "Pliva" Pharmaceutical and Chemical Works, Zagreb, Croatia, Yugoslavia. 2. Member of the Editorial Board, "Croatica chemica acta, Arhiv za kemiju" (for Kukolja).

42 / H

S/081/62/000/021/019/069 B156/B101

AUTHORS: Kukolja, S., Polak, Lj., Krnjević, H., Videk, K.

1.

TITLE: Substances acting on the central nervous system. IV. Derivatives of 2-ethyl-2-phenyl butyramide

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 21, 1962, 157, abstract 21Zh125 (Croat. chem. acta, v. 33, no. 3, 1961, 121 - 126 (Eng.; aummary in Serb.-Croat.)

TEXT: Research to find substances acting on the central nervous system has involved the synthesis of a number of derivatives of $C_6H_5C(C_2H_5)_2COOH_1(acid\ I)$: $C_6H_5C(C_2H_5)_2CONHE$ (II), $C_6H_5C(C_2H_5)_2CONHCOR$ (III) and $4-RC_6H_4C(C_2H_5)COR^4$ (IV). To 0.1 mole I in 200 ml absolute C_6H_6 40 ml SOCl₂ are added; the whole is boiled for 2 hrs, and the volatiles distilles off; without any further purification, the acid chloride is dissolved in 50 ml C_6H_6 or C_6H_6 HCON(CH₃)₂, 0.05 mole of anhydrous C_6H_6 and 0.1 mole of the appropriate amide added, and the mixture heated for 2 hrs at C_6H_6 100°C and left for C_6H_6 1/5

Substances acting on the central ...

s/081/62/000/021/019/069 B156/B101

from the combined filtrates \Re , the gross formula, the yield in \Re , and the melting point in \Re (from alcohol) are given; $\operatorname{C}_2\operatorname{H}_5$, $\operatorname{C}_1\operatorname{H}_2\operatorname{H}_0$, \Re , and the melting point in \Re (from alcohol) are given); $\operatorname{C}_2\operatorname{H}_5$, $\operatorname{C}_1\operatorname{H}_2\operatorname{H}_0$, \Re , \Re , \Re , \Re , $\operatorname{H}_3\operatorname{H}_3\operatorname{H}_0$, $\operatorname{H}_3\operatorname{H}_3\operatorname{H}_3\operatorname{H}_0$, $\operatorname{H}_3\operatorname{H}_3\operatorname{H}_3\operatorname{H}_3\operatorname{H}_3\operatorname{H}_3$, $\operatorname{H}_3\operatorname{H}$

Substances acting on the central ...

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the percentage yield, and the melting point in °C (from alcohol) are given]: CH_3 , $C_{14}H_{19}NO_2$, 18, 89 - 92; $C_{2}H_5$, $C_{15}H_{21}NO_2$, 15, 100 - 102; $C_{6}H_5$, $C_{19}H_{21}NO_2$, 12, 123 - 125; $CHBrCH(CH_3)_2$, $C_{17}H_{24}BrNO_2$, 10, 114 - 116. During 20 min, 4 ml of fuming HNO_3 are added under cooling to 10 g I in 40 ml of concentrated H_2SO_4 , and the mixture is held at 0 - 10°C for 30 min; it is poured onto ice, and 46 % of IV (R = NO_2 , R' = OH) (IVa), $C_{12}H_{15}NO_4$, m.p. 144 - 146°C (from benzene) are separated by recrystallization. 2.4 g IVa and 0.5 g anhydrous Na_2CO_3 in 15 ml water are hydrogenated over 0.05 g of Pd/C at $\sim 2CO^2$ and 760 mm; the filtrate is neutralized with HCl, and 70 % of IV (R = NH_2 , R' = OH) (IVb), $C_{12}H_{17}NO_2$, m.p. 166 - 167°C (from alcohol), are separated. 1 g IVb in 10 ml ($CH_3CO)_2O$ is boiled for 2 hrs, the excess of anhydride evaporated, and the residue dissolved in 10 % Na_2CO_3 ; acidifying the alkaline solution provides 41.5% of IV (R = CH_3CONH , R' = OH) (IVc), $C_{14}H_{19}NO_3$, m.p. 197 - 198°C (from aqueous

Card 3/5

 Substances acting on the central ...

S/081/62/000/021/019/069 B156/B101

alcohol). Another substance produced from IVa is IV $(R = N(CH_3)_2, R' = OH)$ (IVd), $CH_{14}N_{21}NO_2$, yield 83 %, m.p. 145 - 146°C (from dilute alcohol). 0.5 g IVd are methylated with CH_2N_2 produced from 1 g nitroso-methyl carbamide, and the methyl ester of IVd $[R = N(CH_3)_2, R' = OCH_3]$, $C_{15}H_{23}NO_2$, is obtained; yield 90 %, m.p. 79 - 80°C. The methyl ester $(R = CH_3CONH, R' = OCH_3)$, $C_{15}H_{21}NO_3$, (yield 90 %, m.p. 143 - 144°C) is synthesized in an analogous manner from 0.6 g of IVc. 10 g IV $(R = H, R' = NH_2)$ (IVe) are cooled with ice and added to 50 ml of concentrated H_2SO_4 , and during 20 min at 0 - 10°C 4 ml of fuming HNO_3 are added drop by drop; the mixture is held in ice for 30 min, and poured out onto ice; the resultant product is 54 % of IV $(R = NO_2, R' = NH_2)$ (IVf), $C_{12}H_16N_2O_3$, m.p. 127 - 128°C. To 1 g IVa in 10 ml C_6H_6 2 ml of $SOCl_2$ are added; the mixture is boiled for 2 hrs, the volatile substances evaporated, the residue dissolved in 10 ml C_6H_6 and saturated with NH_3 gas; the product is 40 % of IVf. By nitrating Card 4/5

Substances acting on the central...

S/081/62/000/021/019/069 B156/B101

10 g C₆H₅C(C₂H₅)₂CN (VII) in a manner analogous with the production of IVa, 47.5 % of 4-N0₂C₆H₄C(C₂H₅)₂CN (VIII), C₁₂H₁₄N₂O₂, m.p. 81 - 83°C (from benzene) are synthesized. 3 g VIII in 10 ml 96 % H₂SO₄ are heated for 8 hrs at 70°C and poured onto ice; 67 % IVf are extracted with C₆H₆. IV is correspondingly produced in a manner analogous with that described above for the synthesis of IVd and IVe (R, R', the gross formula, the given); N(CH₃)₂, NH₂, C₁₄H₂₂N₂O, IVf, 79, 119 - 120; NH₂, NH₂, C₁₂H₁₈N₂O, IVf, 53, 142 - 143. 85 g VII, 250 ml concentrated H₂SO₄, and 25 ml water are heated at ~100°C for four hrs, and then after cooling poured onto ice, C₆H₆ being used for extracting 81 % of nonpurified IVe, m.p. 49 - 51°C. IV g of nonpurified IVe are treated by the method described earlier (see CH₃COOH, and 90 % I, m.p. of 90 - 91°C, is produced. For communication III, see RZhKhim, 1962, 8Zh134. [Abstracter's note: Complete translation.]

SZUCS, Sandor, dr.; NYIREDY, Geza, dr.; NAGY, Piroska, dr.; VIDEKI, Karoly, dr.

Bronchogenic cysts of the mediastinum. Tuberkulosis 1/ no 8:24/ 2/2

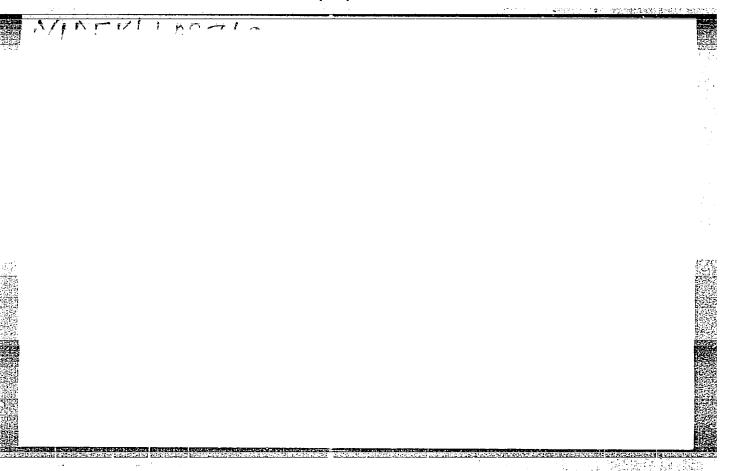
Bronchogenic cysts of the mediastinum. Tuberkulozis 14 no.8:246-249 Ag '61.

1. A Budapesti Orvostudomanyi Egyetem Tudogyogyaszati Klinikaja (Igaz-gato: Kovats Ferenc dr. egyetemi tanar, az orvostudomanyok doktora)

(MEDIASTINUM dis) (BRONCHI dis) (CYSTS)

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VIDEKI, Laszlo; BONTOVITS, Lajos

Inner content losses in storing tomatoes. Elelm ipar 11 no.3/4: 106-108 Je-Jl '57.

1. Duna-Tiszakozi Mezogazdasagi Kiserleti Intezet, Kecskemet.

ji